

Crystal structure and Hirshfeld surface analyses of copper(I) complexes with thiosemicarbazides

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Thiosemicarbazides are organic compounds that have been mainly used as precursors for the synthesis of biologically active heterocycles [1]. The use of these compounds as chelating agents has attracted interest due to the possibility of obtaining metal complexes with pronounced biological, luminescent and catalytic properties [2–4]. Aiming to obtain metal complexes with pharmacological potential, herein we report the synthesis and structural analysis of two copper(I) complexes with the isomeric thiosemicarbazides H_2L^{nic} and H_2L^{iso} as ligands (Figure 1).

The single crystal X-ray diffraction results reveal the formation of the cationic and isostructural complexes $[Cu(H_2L^{nic})(PPh_3)_2]^+$ and $[Cu(H_2L^{iso})(PPh_3)_2]^+$, which have a nitrate as counterion. Both complexes are formed by a copper(I) cation coordinated to two triphenylphosphine ligands and one bidentate and zwitterionic thiosemicarbazide, being inserted into a distorted tetrahedral geometry. The difference in the position of the nitrogen atom in the structure of the ligand enabled the obtaining of different crystal structures, with the complex $[Cu(H_2L^{nic})(PPh_3)_2]NO_3$ crystallizing as an anhydrous complex, while $[Cu(H_2L^{iso})(PPh_3)_2]NO_3$ was obtained as a methanolic solvate. The Hirshfeld surface analysis was performed to evaluate the intermolecular interactions in the two structures, with different patterns of non-covalent interactions being observed for each structure.

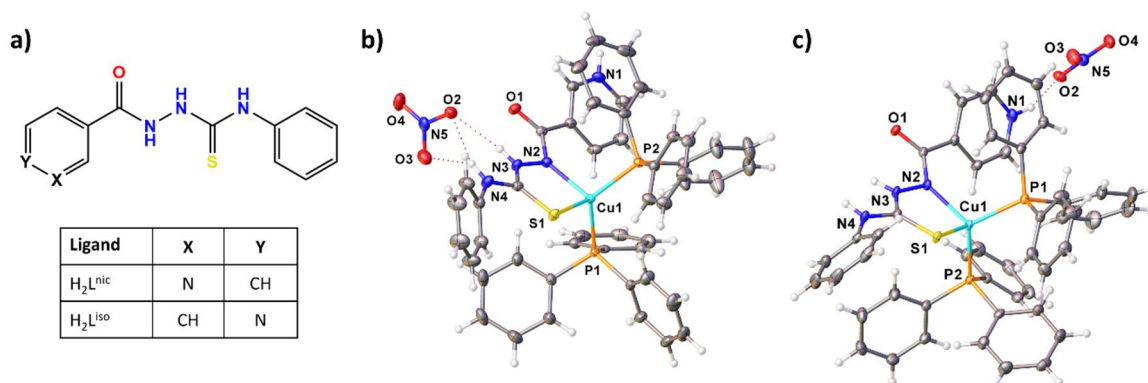


Figure 1. (a) Representation of the thiosemicarbazides and ORTEP type illustration of the complexes (b) $[Cu(H_2L^{nic})(PPh_3)_2]NO_3$ and (c) $[Cu(H_2L^{iso})(PPh_3)_2]NO_3$, with ellipsoid drawn at 50% of probability.

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