

Structural analysis of the composite based in TiO₂ and carbon dot from *o*-phenylenediamine

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TiO₂ nanoparticles when are modified with carbon dots, some significant improvements can be achieved mainly in terms of broadening the light absorption wavelengths [1]. This work aims to obtain two nanocomposites with different percentages of *o*-phenylenediamine (*o*-OPDA) carbon dot (CDsOPDA) adsorbed on the surface of rutile titanium dioxide (TiO₂) for photodegradation of dyes. The studies vise to assess the photocatalytic efficiency of the materials. In the synthesis, the nanocomposite (1) consisted in 400 mg TiO₂ and 20 mg CDsOPDA and nanocomposite (2) 400 mg TiO₂ and 200 mg CDsOPDA. The proportion of TiO₂ and *o*-OPDA were dispersed 50% ethanol: water and added in a hydrothermal reactor. Nanocomposites were synthesized at 473 K in a muffle and posteriorly purified under successive washing with water. The XRD analysis did not reveal any change in the crystal structure of rutile TiO₂ between nanocomposites. These results are indicative that the carbon dot is adsorbed on the surface of the semiconductor and is independent of *o*-OPDA concentration in the preparation. The isotherms obtained in ASAP suggest that the nanocomposites are non-porous adsorbents and macroporous. The pore diameters and volumes of the rutile TiO₂ do not present significant differences with the nanocomposites. However, was observed a decrease in the surface area (BET) for nanocomposite (1) in comparison with (2). The reflectance spectra (Figure 1A) of the nanocomposite (2) confirmed its absorption in the visible-light region and a displacement of the band gap in compared to TiO₂. Photocatalysis studies demonstrated the potential application of nanocomposite (1) in dye degradation red light (Figure 1B) [2].

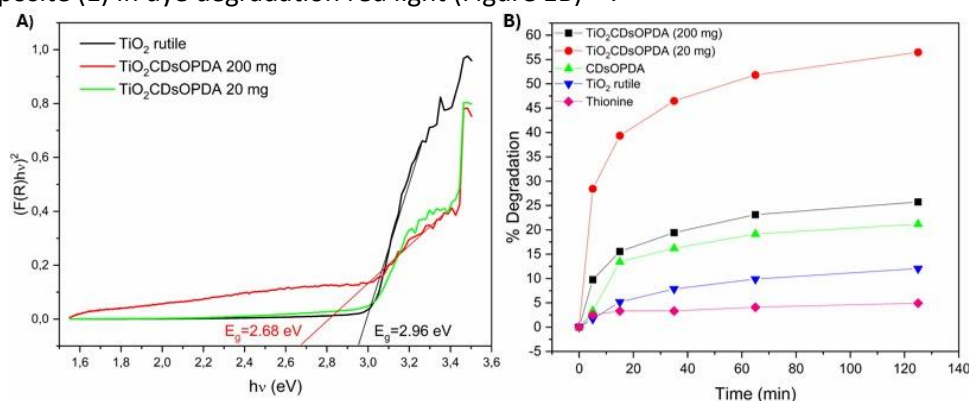


Figure 1. Reflectance spectra (A). %Thionine Photodegradation in function of the time irradiating with 630 nm light in different times (B).

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References

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