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Structural and magnetic characterization of a new oxalate complex of ruthenium(III)

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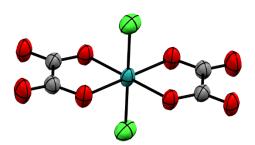
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Ruthenium(III) coordination compounds have been of great interest in several fields, including molecular magnetism due to the fact that 4d-based building blocks can be used for constructing intriguing polynuclear magnetic molecules ^{1,2}. In this work, trans-[PPh₄]₂[RuCl₂(ox)(Hox)]-5H₂O (ox = oxalate) has been obtained in a one-pot microwave-assisted hydrothermal synthesis, directly from RuCl₃·xH₂O, oxalic acid and potassium oxalate. In DMF solution, a quasi-reversible redox couple is observed with E^o = 340 mV vs Ag/AgCl. Single-crystal XRD shows that the compound crystallizes in a triclinic P-1 space group, where the [Ru(ox)(Hox)]²⁻ anions are located in the four vertices and the centre of the unit cell. These anions and the water molecules of crystallization are arranged in parallel planes, layered by the PPh₄+ cations, the entire structure being stabilized mainly by hydrogen bonds. The minimum distance between metallic centres is 9.77 Å. Static magnetic measurements show that $\chi_M T$ at room temperature is equal to 0.73 cm³ K mol⁻¹, and upon cooling this value remains practically constant until 8 K. This is in good agreement with the expected behavior for an isolated ruthenium(III) mononuclear complex. Dynamic magnetic measurements reveal field-induced slow relaxation of magnetization below 8.5 K under a dc field of 1 kOe. The relaxation turns out to be a combination of Raman and direct processes.



ORTEP-type drawing of [RuCl₂(Hox)(ox)]²⁻ with thermal ellipsoids plotted at 50% probability level

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