





Belo Horizonte, September 12 - 15th 2024

New Re(I)-modified Covalent Organic Frameworks for CO₂ photoreduction

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Thematic Area: Photochemistry

Keywords: Covalent Organic Framework, Re(I) carbonyl complex, artificial photosynthesis.

Recently, great research efforts have been spent to realize solar energy conversion by mimicking natural photosynthesis. This work aims to introduce two new covalent organic framework (COF) modified with Re(I) tricarbonyl complexes, ETTA-BPY-Re(CO)₃Cl and DBC-BPY-Re(CO)₃Cl (Figure 1), to explore their potential as catalysts for CO2 photoreduction. The pristine COFs were synthesized from 4,4',4"'-(ethene-1,1,2,2-tetrayl)tetraaniline (ETTA), with 2,2'-bipyridine-5,5'-dicarbaldehyde (BPY) and dibenzo[g,p]chrysene-2,7,10,15-tetraamine (DBC) with BPY using 1,4-Dioxane/Mesitylene (1:1) and acetic acid (10%) for 3 days at 120 °C. Modification with Re(CO)₅Cl occurred in toluene under Ar. Both materials underwent initial characterization by PXRD, SEM, N₂ sorption and FTIR spectroscopy confirming the proposed structure. UV-Vis spectroscopy and PL measurements confirmed the electronic interaction between the orbitals of the COF and the Re(I) center. Photocatalytic tests (λ_{irr} > 370 nm; P_{irr} = 100 mW cm⁻²) were carried out in acetonitrile using 1,3-dimethyl-2-phenyl-2,3-dihydro-1H-benzo[d]imidazole (BIH) as sacrificial agent. The pristine COFs showed no activity for CO2 photoreduction. ETTA-BPY-Re(CO)₃Cl showed a maximum conversion rate of 135 μmol g⁻¹ h⁻¹, while the DBC-BPY-Re(CO)₃Cl showed a maximum rate of 1168 μmol g⁻¹ h⁻¹, which is the highest CO evolution rate for these Re(I) modified COFs in the literature [1]. This highlights the importance of the selection of the building blocks for COF-based CO2 photocatalysts. DBC, with more unsaturated bonds, possibly acts more efficiently in injecting and collecting light than the ETTA counterpart. Mechanistic studies as well as optimization of the reaction conditions are current in progress aiming at the enhancement of the performance of the new catalyst.

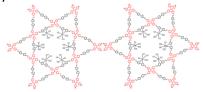


Figure 1. Structures of ETTA-BPY-Re(CO)₃Cl (left) and DBC-BPY-Re(CO)₃Cl (right).

Acknowledgments: The authors are thankfull to Fundação de Amparo Pesquisa do Estado de Minas Gerais, Conselho Nacional de Desenvolvimento Científico e Tecnológico and Coordenação de Aperfeiçoamento de Pessoal de Nível Superior.

References

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