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## High symmetry and unusually long lifetimes in RE<sup>3+</sup>-doped Y<sub>2</sub>Sn<sub>2</sub>O<sub>7</sub> materials

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Pyrochlore compounds with the general formula A2B2O7 have been explored as promising host materials for rare earth (RE3+) ions. These materials crystallize in a pyrochlore structure, where trivalent cations (A<sup>3+</sup>) and tetravalent cations (B<sup>4+</sup>) alternate in a Face Centered Cubic (FCC) lattice with vacant anion sites ensuring charge neutrality. Both A<sup>3+</sup> and B<sup>4+</sup> ions exhibit D<sub>3d</sub> symmetry [1]. RE<sup>3+</sup> can be easily incorporated into this lattice due to their similar ionic radii with A3+ cations, assuming the same symmetry. In this sense, a detailed investigation of  $RE^{3+}$ -doped  $Y_2Sn_2O_7(RE^{3+} = Eu^{3+}, Er^{3+} \text{ and } Yb^{3+})$ powder prepared by the coprecipitation method followed by annealing at higher temperatures (700 to 1300 °C) is presented. This work explores the stabilization of the crystalline phase of Y<sub>2</sub>Sn<sub>2</sub>O<sub>7</sub> with changes in annealing temperature, monitoring structural changes through X-ray diffraction and Raman spectroscopy. Additionally, static and dynamic measurements of RE<sup>3+</sup> luminescence are investigated, elucidating not only the crystalline structure but also the distribution of active ions. X-ray diffraction of the RE<sup>3+</sup>-doped Y<sub>2</sub>Sn<sub>2</sub>O<sub>7</sub> shows the crystallization of pyrochlore phase, with space group Fd3m, and cassiterite SnO<sub>2</sub> as a secondary phase. The stabilization of the crystalline phase of interest was confirmed via Rietveld refinement and Raman spectroscopy, indicating higher crystallinity at higher temperatures. The Eu<sup>3+</sup> emission spectra were obtained under excitation at different wavelength, at the charge transfer (CT) band (~260nm), and at the electronic transitions  ${}^{7}F_{0} \rightarrow {}^{5}L_{6}$  (394nm)  ${}^{7}F_{0} \rightarrow {}^{5}D_{2}$ (464nm) and  ${}^{7}F_{0} \rightarrow {}^{5}D_{1}$  (525nm), where the  ${}^{5}D_{0} \rightarrow {}^{7}F_{J}$  (J = 0, 1, 2, 3 and 4) transitions could be observed. These spectra showed the presence of Eu<sup>3+</sup> ions in a high-symmetry site, confirming their incorporation into the lattice [2]. For these samples, the measured lifetimes values exceeded 5 ms depending on the wavelength. For the samples codoped with Er<sup>3+</sup> Yb<sup>3+</sup> ions, the emission spectra were obtained under excitation at 980 nm attributed at the electronic transitions  ${}^4I_{11/2} \rightarrow {}^4I_{15/2}$  of Er<sup>3+</sup> ions or  ${}^2F_{5/2} \rightarrow {}^2F_{7/2}$  of Yb<sup>3+</sup>. The lifetime values from the excited state <sup>4</sup>I<sub>13/2</sub> (Er<sup>3+</sup>) for the samples annealed at 1300°C were higher than 20 ms. Intense upconversion luminescence was observed under 980 nm, affording a yellowish-green emission that can be seen by the naked eyes. The long lifetimes values observed in these samples suggest that these materials are suitable for in time-resolved analysis, including bioimaging techniques [3].

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## References

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