

Lessons on magnetic anisotropy from lanthanide-chalcogen complexes

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Lanthanide complexes exhibit high magnetic anisotropy, which can be tuned by appropriate molecular design. In the present work, dichalcogenoimidodiphosphinate ligands, $N(EPPH_2)_2^-$ ($E = O, S$, and Se),^[1] produced six crystalline $[Ln\{N(EPPH_2)_2\}_3]$ complexes ($Ln = Dy$ or Er , Fig. 1a) characterized by structural, spectroscopic, and magnetometric analyses. Changing the chalcogen donor atoms led to significant structural changes in the resulting complexes, with a relevant impact on magnetic properties. Six and nine-coordinate Ln^{3+} ions were observed with the lighter chalcogens (O and S , respectively), while Se gave rise to seven-coordinate products. For the former, the short Ln-chalcogen bond distances and crowded coordination spheres led to fast magnetic relaxation due to high rhombicity, confirmed by *ab initio* calculations. Conversely, in $[Dy\{N(SePPH_2)_2\}_3]$, the additional coordination of one nitrogen atom led to a uniquely favorable Dy–N axis corresponding to the easy anisotropy axis of the structure. *Ab initio* calculations revealed for this complex a perfectly axial ground state mainly composed by the highest m_J projection ($m_J = \pm 15/2$; $g \approx [0, 0, 20]$) and a first excited state at higher energy than in all other complexes in the series (Fig. 1b). AC magnetic susceptibility measurements confirmed the theoretical findings, revealing slow magnetic relaxation for $[Dy\{N(SePPH_2)_2\}_3]$ in both zero and applied static fields. These results highlight the importance of tailored syntheses in fine-tuning magnetic properties.

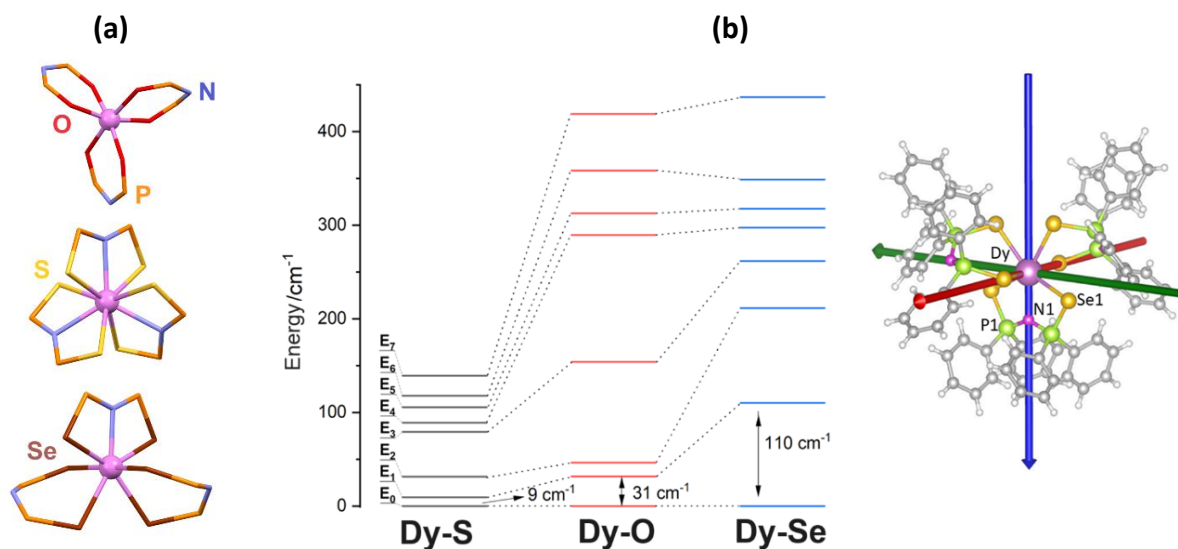


Figure 1: (a) Representation of the $[Ln\{N(EPPH_2)_2\}_3]$ coordination geometries. (b) Left: Calculated energies of the $^6H_{15/2}$ Kramer's doublets (E_n) in the Dy^{3+} complexes. Right: Structure of $[Dy\{N(SePPH_2)_2\}_3]$, emphasizing the easy magnetization axis in the ground state (blue).

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